Optical Petrology

Use of the petrographic microscope

Modified by James Wray from:

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Why use the microscope??



- Identify minerals
- Determine rock type
- Determine crystallization sequence
- Observe frozen-in reactions
- Document deformation history
- Constrain P-T history
- Note weathering/alteration

The petrographic microscope



Also called a polarizing microscope

In order to use the scope, we need to understand a little about the **physics of light**

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What happens as light moves through the scope?



Light beam = numerous photons, each vibrating in a different plane

Vibration in all directions ~ perpendicular to propagation direction

Unpolarized Light

1) Light passes through the lower polarizer



Only the component of light vibrating in E-W direction can pass through lower polarizer – light intensity decreases





Conclusion has to be that minerals somehow **reorient** the planes in which light is vibrating; some light passes through the upper polarizer



But, note that some minerals are better at this than others (i.e., some grains stay dark and thus can't be reorienting light)

Some generalizations and vocabulary

- Amorphous materials and isometric minerals (e.g., garnet) are isotropic – they cannot reorient light. These minerals are always extinct in crossed polars (XPL).
- All other minerals are anisotropic they are all capable of reorienting light.
- All anisotropic minerals contain one or two special *propagation* directions that do **not** reorient light.
 - ▲ Minerals with one special direction are called uniaxial (hexagonal, tetragonal)
 - ▲ Minerals with two special directions are called biaxial (orthorhombic, monoclinic, triclinic)

Rotating the stage

Anisotropic minerals change color as the stage is rotated; these grains go black 4 times in 360° rotationexactly every 90°



Transmission by the Analyzer

Determined by:

- a) Angle between analyzer and polarizer (fixed at 90°)
- b) Angle between polarizer and closest privileged direction of xl
 - ▲ When polarizer || either privileged vibration direction → extinct, since only one ray & it's cancelled
 - Every crystal goes extinct 4 times in 360° rotation (unless isotropic)

Pleochroism

 Changes in absorption color in PPL as rotate stage (common in biotite, amphibole...)



Biotite as stage is rotated

Mineral properties in PPL: relief

- Relief is a measure of the relative difference in n between a mineral grain and its surroundings
- Relief is determined visually, in PPL
- Relief is used to estimate n

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garnet:	n = 1.72-1.89
quartz:	n = 1.54-1.55
epoxy:	n = 1.54



Quartz has low relief

Garnet has high relief

Mineral properties in PPL: relief

- Relief is a measure of the **relative** difference in **n** between a mineral grain and its surroundings
- Relief is determined visually, in PPL
- Relief is used to estimate n



Olivine has high reliefPlagioclase has low relief

olivine:	n = 1.64-1.88
plag:	n = 1.53-1.57
epoxy:	n = 1.54

What causes relief?

Difference in speed of light (n) in different materials causes refraction of light rays, which can lead to focusing or defocusing of grain edges relative to their surroundings



Birefringence (in lunar basalt)

Plane-polarized ("normal") illumination



Cross-polarized view





A few additional properties of individual minerals

Cleavage - number and orientation of cleavage planes

Twinning - type of twinning, orientation

Extinction angle - parallel or inclined? Angle?

Habit - characteristic form of mineral

Cleavage

Most easily observed in PPL (upper polarizer out), but visible in XN as well

- No cleavages: quartz, olivine
- 1 good cleavage: micas
- 2 good cleavages: pyroxenes, amphiboles









2 cleavages intersecting at 60°/120°: amphibole





random fractures, no cleavage: olivine



Presence and style of twinning can be diagnostic



Twins are usually most obvious in XN (upper polarizer in)

Twinning - some examples



Clinopyroxene (augite)

• Simple twin on {100}



Plagioclase

- Simple (Carlsbad) twinning
- Polysynthetic albite twinning
- Pericline twinning

Extinction angle - parallel extinction

- All uniaxial minerals show parallel extinction
- Orthorhombic minerals show parallel extinction

(this is because the crystal axes and indicatrix axes coincide)

orthopyroxene





Extinction angle - inclined extinction

Monoclinic and triclinic minerals have inclined extinction

(and extinction angle helps to identify them)





clinopyroxene

Habit or form

acicular anhedral/irregular bladed blocky elongate euhedral fibrous prismatic rounded tabular



Review - techniques for identifying unknown minerals

Start in PPL:

- Color/pleochroism
- Relief
- Cleavages
- Habit





Then go to XPL:

- Birefringence
- Twinning
- Extinction angle
- Uniaxial or biaxial?
- Additional info from accessory plates

Chapter 3: Igneous Textures

Figure 3.1. Idealized rates of crystal nucleation and growth as a function of temperature below the melting point. Slow cooling results in only minor undercooling (T_a) , so that rapid growth and slow nucleation produce fewer coarse-grained crystals. Rapid cooling permits more undercooling (T_b) , so that slower growth and rapid nucleation produce many fine-grained crystals. Very rapid cooling involves little if any nucleation or growth (T_c) producing a glass.



Figure 3.2. Backscattered electron image of quenched "blue glassy pahoehoe," 1996 Kalapana flow, Hawaii. Black minerals are felsic plagioclase and gray ones are mafics. a. Large embayed olivine phenocryst with smaller plagioclase laths and clusters of feathery augite nucleating on plagioclase. Magnification ca. 400X. **b**. ca. 2000X magnification of feathery quenched augite crystals nucleating on plagioclase (black) and growing in a dendritic form outward. Augite nucleates on plagioclase rather than pre-existing augite phenocrysts, perhaps due to local enrichment in mafic components as plagioclase depletes the adjacent liquid in Ca, Al, and Si. © John Winter and Prentice Hall.



Figure 3.3. a. Volume of liquid (green) available to an edge or corner of a crystal is greater than for a side. b. Volume of liquid available to the narrow end of a slender crystal is even greater. After Shelley (1993). Igneous and Metamorphic Rocks Under the Microscope. © Chapman and Hall. London.





Figure 3.4. a. Skeletal olivine phenocryst with rapid growth at edges enveloping melt at ends. Taupo, N.Z. **b.** "**Swallow-tail**" plagioclase in trachyte, Remarkable Dike, N.Z. Length of both fields ca. 0.2 mm. From Shelley (1993). Igneous and Metamorphic Rocks Under the Microscope. © Chapman and Hall. London.

Figure 3.5. a. Compositionally zoned hornblende phenocryst with pronounced color variation visible in plane-polarized light. Field width 1 mm. **b.** Zoned plagioclase twinned on the carlsbad law. Andesite, Crater Lake, OR. Field width 0.3 mm. © John Winter and Prentice Hall.





Figure 3.6. Examples of plagioclase zoning profiles determined by microprobe point traverses.

a. Repeated sharp reversals attributed to magma mixing, followed by normal cooling increments.

b. Smaller and irregular oscillations caused by local disequilibrium crystallization.

c. Complex oscillations due to combinations of magma mixing and local disequilibrium.

From Shelley (1993). Igneous and Metamorphic Rocks Under the Microscope. © Chapman and Hall. London.



Figure 3.7. Euhedral early pyroxene with late **interstitial** plagioclase (horizontal twins). Stillwater complex, Montana. Field width 5 mm. © John Winter and Prentice Hall.



Figure 3.8. Ophitic texture. A single pyroxene envelops several well-developed plagioclase laths. Width 1 mm. Skaergård intrusion, E. Greenland. © John Winter and Prentice Hall.

Figure 3.10. Olivine mantled by orthopyroxene



(**a**) plane-polarized light

(b) crossed nicols: olivine is extinct and the pyroxenes stand out clearly.



Basaltic andesite, Mt. McLaughlin, Oregon.Width ~ 5 mm.© John Winter and Prentice Hall.



Figure 3.13. Flow banding in andesite. Mt. Rainier, WA. © John Winter and Prentice Hall.

Figure 3.12a. Trachytic texture in which microphenocrysts of plagioclase are aligned due to flow. Note flow around phenocryst (P). Trachyte, Germany. Width 1 mm. From MacKenzie *et al.* (1982). © John Winter and Prentice Hall.



Figure 3.18. (c-d) Tartan twins in microcline. Field widths ~1 mm. © John Winter and Prentice Hall.





Figure 3.20. a. Pyroxene largely replaced by hornblende. Some pyroxene remains as light areas (Pyx) in the hornblende core. Width 1 mm. b. Chlorite (green) replaces biotite (dark brown) at the rim and along cleavages. Tonalite. San Diego, CA. Width 0.3 mm. © John Winter and Prentice Hall.







Figure 3.21. **Myrmekite** (dendritic quartz) formed in plagioclase at the boundary with K-feldspar. Photographs courtesy © L. Collins. http://www.csun.edu/~vcgeo005